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q= m? c?"7: identify each variables by name & the units associated with it. 2. Heat is not the same as temperature, yet they are related. Explain how they differ from each other. a. Perform calculations using: (q= m?c?"7±b. Deter PLQHLILW-VH QGRWKHUPLFRUHI Page 15/39

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Izess Funicello oduction to Specific_ Heat WKST.pdf ... Worksheet-Introduction to Specific Heat Capacities Heating substances in the sun: The following table shows the temperature after 10.0 a of 4 different Page 16/39

substances have been in direct sunlight for up to 60 minutes. Time (minutes) Air (° C) Water (° C) Sand (° C) Metal (° C) O (initial) 25°C 25°C 25°C 25°C

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calculations usin 1. Gold has a specific heat of 0.129 J/(gx0C). How

Answers

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J/(gxOC) Endo

thermic or 5. Copper has a specific heat of 0.385 J/(gx0C). A piece of copper absorbs 5000 J of energy and undergoes a temperature change from 100 oc to 200 0 C.

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joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron. Specific Heat Worksheet

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