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stoichiometry When the reactants are present in the correct amounts, the reaction will produce products. What happens if there are more or less of some of the reactants present?

[Basic Stoichometry](#)

Learn the basics of the Phet lab and worksheet

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Create your own sandwich and then see how many sandwiches you can make with different amounts of ingredients. Do the same with chemical reactions. See how many products you can make with different amounts of reactants. Play a game to test your understanding of reactants, products and leftovers. Can you get a perfect score on each level?

[Reactants, Products and Leftovers - PhET](#)

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Phet Lab Answer Key Acid And Base Phet Lab Answer Key Acid Thank you for downloading Phet Lab Answer Key Acid And Base. Introduction/ Purpose: Stoichiometry is the study of the quantitative. The substance is originally a gas at 160 ° C, condenses at 132 ° C, and freezes at 67 ° C.

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Stoichiometry Review Assignment Answer Key Example 1: Calculate the mass of a magnesium, Mg, atoms in grams. 24.035 g Mg , 1 mol Mg , 1 molecule Mg = 4.04 x 10-23 g/ Mg atom 1 mol Mg 6.02 x 1023 molecules 1 atom Mg Example 2: Calculate the number of atoms in one-millionth of a gram of magnesium, Mg.

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