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Bonding Pt 1

Part 8 Covalent

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drawing water
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240 Chapter 8 □

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Covalent
Bond-!).)DEAAtoms
gain stability when
they share
electrons and form
covalent bonds.
Real-World Reading
Link Have you ever
run in a three-
legged race? Each
person in the race
shares one of their
legs with a

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teammate to form
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Chapter 8 Covalent Bonding and Molecular Structure

8-3 There are two types of repulsive forces between the two atoms. First, the nuclei repel because they are both positively charged. Second, the electrons repel because they are

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both negatively charged. The attractive forces between the two atoms result from the

Chapter 8:
Covalent Bonding
and Molecular
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Section 8.4 – Polar
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Covalent bonds involve sharing electrons between atoms. When the atoms in the bond pull equally, the bonding electrons are shared equally, and the bond is nonpolar. When the atoms in the bond pull unequally, the bonding electrons are pulled closer to

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one atom, and the
bond is polar.

Chapter 8 –
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Covalent bond A
bond formed by
the sharing of
electrons between
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8.1 and 8.2

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Define a polar
covalent bond. 7.

When atoms form a
polar covalent
bond, the atom
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attraction for
electrons has a
partial charge.

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Covalent Bond

Single Covalent

Bonds (cont.) □

Atoms in group 16

can share two

electrons and form

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two covalent bonds. □ Water is formed from one oxygen with two hydrogen atoms covalently bonded to it.

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a rule that usually
tells how many
electrons are
shared to form a
covalent bond □
Describe how
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□ Predict when two
atoms are

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