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6.2 6.2.1 Vinegar is a  
dilute form of acetic acid.  
A sample of acetic acid has  
the following percentage  
composition: 39,9% carbon

6,7% hydrogen 53,4% oxygen

Determine the molecular  
formula of acetic acid if  
the molar mass of acetic  
acid is  $60 \text{ g}\cdot\text{mol}^{-1}$ . (6)

6.2.2 Give a reason why  
acetic acid is regarded as a  
monoprotic acid. (2)

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