

# Periodic Trends Pogil

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Periodic Trends . 6. Locate the numbers in Model 1 that represent the ionization  
energy. The ionization energy is the amount of energy needed to remove an electron  
from an atom. a. Using your knowledge of Coulombic attraction, explain why  
ionization—removing an

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Across the periodic table, sizes of atoms show the following trends, with many  
irregularities: Size increases down a group. The outermost electrons are in  
successively more extensive orbitals as  $n$  increases. Size decreases across a period.  
Electrons are added to the same shell and do not shield one

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Periodic Trends POGIL Model 1 – Main Group Elements. The periodic table is often considered to be the “best friend” of chemists and chemistry students alike. ... Support your answer, using examples from two periods.

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Periodic trends are specific patterns that are present in the periodic table that illustrate different aspects of a certain element, including its size and its electronic properties. Major periodic trends include: electronegativity , ionization energy , electron affinity , atomic radius , melting point, and metallic character .

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performance will be judged based on achievement of a target, he ' ll work harder to attain it.

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The trends in these properties as you go across periods and down groups is the subject of this Chem POGIL Success Criteria:

- Understand the meaning of atomic radius, reactivity, electronegativity, ionization energy
- Recognize trends in atomic radius, reactivity, ionization energy and electronegativity as you go across periods and down groups.

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