Numericals Chemistry Chapter Solid State

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Rohit Dahiya
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ALL NUMERICALS Solid state, NCERT BOOK SOLUTIONS ( EXERCISE QUESTION
1.1 TO 1.26), CLASS -12 Numericals Chemistry Chapter Solid State
Numericals Chemistry Chapter Solid State (v) 13-15 group compounds :
When the solid state materials are produced by combination of
elements of groups 13 and 15, the compounds thus obtained are called
13-15 compounds.
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Numericals Chemistry Chapter Solid State
The post Solid State Chemistry Numericals contain all important
questions related to the topic Que 1. A solid has a cubic structure
in which X atoms are located at the corners of the cube, Y atoms are
at the cube centers and O atoms are at the edge centers. What is the
formula of the compound?

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Solid state numericals

(v) 13-15 group compounds: When the solid state materials are produced by combination of elements of groups 13 and 15, the compounds thus obtained are called 13-15 compounds. For example, InSb, AlP, GaAs, etc. 12-16 group compounds: Combination of elements of groups 12 and 16 yield some solid compounds which are referred to as 12-16 compounds. For example, ZnS, CdS, CdSe, HgTe, etc.

NCERT Solutions For Class 12 Chemistry Chapter 1 The Solid ... Chemistry Notes for class 12 Chapter 1 The Solid State Solids Solids are the chemical substances which are characterised by define shape and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules or ions) are closely packed and held together by strong interparticle forces Types of Solids

Chemistry Notes for class 12 Chapter 1 The Solid State
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Class 12 Chemistry Chapter 1 To discover more about this chapter,
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CBSE Class 12 Chemistry Notes Chapter 1 The Solid State
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Solid states deal with arrangements of particles in solids that
result in several types of structure in solids. These differences in
the structural unit also result in different properties of each SolidState.

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NCERT Exemplar Class 12 Chemistry Chapter 1 Solid State ...

Ans: The main concepts under this chapter are: Characteristics and properties of the solid-state Amorphous and Crystalline solids Classification of crystalline solids Crystal Lattices and unit cells Unit cells Types of unit cells Crystal lattices Characteristics of a crystal lattice Number of atoms ...

Class 12 Chemistry Revision Notes for Chapter 1 - The ...

Numericals Chemistry Chapter Solid State (v) 13-15 group compounds:
When the solid state materials are produced by combination of elements of groups 13 and 15, the compounds thus obtained are called 13-15 compounds. For example, InSb, AlP, GaAs, etc. 12-16 group compounds: Combination of elements of groups 12 and 16 yield some solid

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Answer: (i) Those solids which have 3-D arrangement of atoms, ions or molecules are called crystalline solids, e.g. diamond. (ii) This is due to shifting of an ion from its normal position to an interstitial, in the crystal lattice, thus causing a vacancy in the original position. The overall density is not affected by this change.

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The Solid State Class 12 Important Questions Long Answer Type (LA)
Question 106. (a) An element has an atomic mass 93 g mol-1 and
density 11.5 g cm-3. If the edge length of its unit cell is 300 pm,
identify the type of unit cell. (b) Write any two differences between
amorphous solids and crystalline solids. (Delhi) (2017) Answer: (a)
Given:

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