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0.98 x (correct answer) 0.95 x (correct answer) 0.90 x (correct answer) 1.02 x
(correct answer) 1.05 x (correct answer) 1.10 x (correct answer) Correct answer:
(actual mass divided by 63.546 times 6.022e23) 4 3 2 Figure 1. Procedure 1.

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0.98 x (correct answer) 0.95 x (correct answer) 0.90 x (correct answer) 1.02 x
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(actual mass divided by 63.546 times 6.022e23) 4 3

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Moles Lab Activity 3: Compounds —Water, Compounds Sodium Chloride, Compounds Chalk Compounds—The Fictitious ompound “Cambium” Moles Lab Activity 4: Solutions—Aqueous Copper (II) Sulfate Pent hydrate, Solutions—Alum Moles Lab Activity 5: Synthesis of an Oxide of Copper

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The fastest way to obtain a relative mass of beans would be to count the beans. The fastest way to obtain a mole of beans would be to weigh them. (At least in principle. The mass of a mole of beans would be incredibly large- on the order of 10²² g.) Part III. All atomic masses agree with the relative masses to three significant figures.

Laboratory Activity 1: Teacher Notes Continued

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Have a conceptual understanding of the mole as the method of “counting” items and finding the mass of items that can’t be seen. 2. Be able to calculate the number of items (molecules, atoms, ions, and formula units) if given the number of moles. 3. Be able to calculate the number of moles if given the number of items.

#20 Introduction to the Mole - Terrific Science

