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Express your answer using a chemical symbol. Hg. Part A Arrange the following oxides in order of increasing acidity. Rank from least acidic to most acidic. To rank items as equivalent, overlap them. CaO, Al₂O₃, SiO₂, CO₂, P₂O₅, SO₃. Part A barium oxide with water Express your answer as a chemical equation. Identify all of the phases in your answer.

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MasteringChemistry

Mastering Chemistry Chapter 2: Rank from most to fewest valence electrons of the following elements: Kr, Ga, Br, As, K. - Answer: Kr (8), Br (7), As (5), Ga (3), K (1). Valence electrons are equal to the atom's main group number. Link: <https://www.coursehero.com/file/20333255/Mastering-Chemistry-Chapter-2/>.

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