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by Millikan Oil
Drop
Experiment

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Plasma Physics
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Charge of an

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Electron: Millikan's
Oil Drop

Experiment

Millikan's oil drop

experiment to

determine charge

of an electron -

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~~atom?~~
Millikan Oil

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*Thomson's Cathode
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Determination
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(Charge of a single electron) Episode 12: The Millikan Experiment - The Mechanical Universe ☐☐

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LABORATORY IV

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MILLIKAN OIL DROP

EXPERIMENT The experiment was devised by

American physicist Robert A. Millikan in 1909. Mil-likan

wanted to determine whether electric charges occurred in

discrete units which were integral multiples of some

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smallest charge e
(which turns out to
be carried by an
electron or proton),
and to measure
that charge.

LABORATORY IV MILLIKAN OIL DROP EXPERIMENT

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Millikan Oil Drop
Experiment In
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Measurement of Q
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adjustments of the
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Drop Experiment LABORATORY IV MILLIKAN OIL DROP EXPERIMENT

Laboratory Iv
Millikan Oil Drop
Laboratory IV:
Millikan Oil Drop
Experiment Eq. 3 is
Stokes Law,
relating the radius
of a spherical body

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to its terminal velocity as it falls in a viscous medium. $F_f = mg$
Stokes' Law, however, becomes incorrect when the velocity of fall of the droplets is less than ~ 0.1 cm/s.

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Experiment

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Chemistry

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EXPERIMENT The

experiment was
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be carried by an
electron or proton),
and to measure
that charge.

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Millikan Oil Drop
Experiment
The Oil Drop
Experiment was
performed by the
American physicist
Robert A Millikan in

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1909 to measure the electric charge carried by an electron. Their original experiment, or any modifications thereof to reach the same goal, are termed as oil drop experiments, in general. Oil Drop Experiment

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Millikan's Oil Drop
Experiment -
Science Facts
Millikan Oil Drop

Lab. In this lab you will be looking for oil drops that can be caught in the electric field between two capacitor plates. Some drops will fall out of your field of view as the

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gravitational force on them is larger than the electric force. Other drops will rise out of your field of view as the gravitational force is too small for the electrical force.

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EXPERIMENT

The experiment was
devised by
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Robert A. Millikan
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occurred in
discrete units
which were integral
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an electric field and then lowering it again. To raise it you apply a constant electric field on the drop that forces it upward. To lower the drop you can either turn off the electric field and just let it fall or you can reverse the

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and to investigate
whether the value
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were carried out
and whether
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how to do
experimental
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nozzle. With the cover plate closed to prevent any loss of oil droplets, the oil droplets then fell through the small hole in the settlement chamber, and again into

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