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Page 4 Mark Scheme: Teachers' version

Syllabus Paper IGCSE - October/November 2011

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International Examinations 2011. 3 (a) Any

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two of: [2] temperature mass / amount of manganese(IV) oxide / volume of manganese(IV) oxide size of manganese dioxide particles allow: pressure ignore: concentration. (b) (i) the greater the concentration the greater the speed / rate increases with concentration [1] ignore: concentration increases speed / more oxygen the grater ...

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Examinations 2011 1 (a) (i) lithium oxide /
strontium oxide [1] (ii) sulfur dioxide /

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nitrogen dioxide [1] (iii) aluminium oxide
[1] (iv) carbon monoxide [1] accept: correct
formulae (b) sulfur dioxide [1] burn (fossil)
fuel containing sulfur / volcanoes [1]

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candidate's 0.26/200 C1 0.0013 e.c.f. from
(a) A1 g/cm^3 B1 [6] 6 (a) (i) reflection OR
wave bounces back M1 from large object/sea
bed A1 (ii) speed = distance/time in any form
C1 1500×0.8 C1

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