

Holt Chemistry Chapter 6 Review

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Holt Chemistry Chapter 6-Review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. bvallejo. The questions that will be on the test. Terms in this set (22) Why is the Diatomic molecule H2 more stable than two separate Hydrogen atom? Each Hydrogen atom now has a stable noble gas electron configuration.

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How it works: Identify the lessons in the Holt Chemistry: Covalent Compounds chapter with which you need help. Find the corresponding video lessons within this companion course chapter.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

[6-Chemical Bonding - Effingham County Schools / Overview](#)

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