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210 Guided Reading and Study Workbook CHAPTER 19, Acids, Bases, and Salts (continued) 8. How do concentrated basic solutions differ from other basic solutions? Brønsted-Lowry Acids and Bases (pages 590-592) 9. How does the Brønsted-Lowry theory define acids and bases? 10. Is the following sentence true or false? Some of the acids and bases ...

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[PDF] Guided Reading and Study Workbook - Free Download PDF 84 Guided Reading and Study Workbook SECTION 9.4 NAMING AND WRITING FORMULAS FOR ACIDS AND BASES (pages 271-273) This section explains the three rules for naming acids and shows how these rules can also be used to write the formulas for acids.

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Honors Chem Unit 12: Acids & Bases Guided Notes 12.1 General Properties and Nomenclature (Guided Reading, Section 18.1 of Textbook) 1. Give two "everyday" examples of acids and two "everyday" examples of bases. 2. General properties of acids: a. Are electrolytes: can_____ electricity. b. Taste _____. c.

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Names and Formulas of Acids Names and formulas of acids are based on an acid's consisting of an anion and enough hydrogen ions to make the acid electrically neutral. The general formula for an acid is $H_n X$, where X is an anion, H is a hydrogen ion, and n is the number of hydrogen ions needed to make the acid neutral.

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Each carbon atom in a lipid's fatty acid chain is joined to another carbon atom by a single bond. Unsaturated A lipid's fatty acids contain more than one double bond. Nucleic Acids (page 47) 18. Nucleic acids contain what kinds of atoms? The monomers that make up nucleic acids are known as. A nucleotide consists of what three parts?

~~Prentice hall Biology Worksheets~~

Study Guide: Acids and Bases 1) List at least three characteristic properties of acids and three of bases. ACIDS pH less than 7 Have a sour taste Change the color of many indicators Are corrosive (react with metals) Neutralize bases Conduct an electric current BASES

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