

Gizmo Density Lab Answers

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To calculate an object ' s density, divide its mass by its volume. If mass is measured in grams and volume in cubic centimeters, the unit of density is grams per cubic centimeter (g/cm³). Calculate the density of each object, and record the answers in the last column of your data table. Label this column “ Density (g/cm³). ”

[Student Exploration: Density Laboratory](#)

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Density Laboratory Gizmo : ExploreLearning

To begin, check that the Type of wave is Transverse, Amplitude is 20.0 cm, Frequency is 0.75 Hz, Tension is 3.0 N, and Density is 1.0 kg/m. (Note: In this Gizmo, “ density ” refers to the linear mass density, or mass per unit length. It is measured in units of kilograms per meter.) Click Play ().

Student Exploration: Waves (ANSWER KEY)

Answers will vary. [An object floats when the weight of the liquid it displaces is equal to the weight of the object. Objects sink when they are heavier than the equivalent volume of liquid. In other words, objects sink when they are denser than the liquid.]

DensityLabSE Key | Buoyancy | Density

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Explore Learning Gizmos: Math & Science Simulations

Gizmo Warm-up: Determining density A mineral is a naturally formed crystal. You can identify a mineral by its properties. In the Mineral Identification Gizmo™, under Choose property to test, select Density.

Student Exploration: Mineral Identification (ANSWER KEY)

One mL is the same volume as one cm³. Density is the amount of mass contained in a given volume. To find the density of an object, divide its mass by its volume. Density is recorded in units of grams per cubic centimeter (g/cm³). Pre Lab (Prior Knowledge) Questions (Do these BEFORE using the Gizmo.)

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laboratory gizmo answer key density laboratory with a scale to measure mass a graduated cylinder to measure volume and a large beaker of liquid to observe flotation the relationship between mass volume density and flotation can be investigated the density of the liquid in the beaker can be adjusted and a variety of objects can be studied during the investigation student exploration phases of ...

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