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Hill would like to ask for A-Level Biology, people often suggest memorizing the mark schemes. I would like to ask, when we do this, do we memorize the question as well and also, do we also memorize from our textbooks? And also, there are a lot of past papers out there, how can we be sure that...

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Alternating Currents Characteristics of alternating currents. Peak current, $I_0 = 3 \text{ A}$. Peak-to-peak current, $I_{p-p} = 6 \text{ A}$. Period, $T = 20 \text{ ms}$. Frequency, $f = 1 / T = 50 \text{ Hz}$. Angular Frequency, $\omega = 2\pi f = 314 \text{ rad s}^{-1}$. Instantaneous current: the current at a particular instant.

Alternating Currents - A-Level Physics ... - XtremePapers

Hey all friends, I have compiled list of pastpapers of Cambridge A Levels for students so they can use these to practice for their exams. Anyway enjoy & share with ...

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A and As Level Geography 9696 Past Papers March, May ...

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A and As Level Economics 9708 Past Papers March, May ...

Electric Fields. Electric field strength / intensity at a point is defined as the force per unit positive charge acting at that point [a vector; Unit: N C⁻¹ or V m⁻¹]. $E = F / q$? $F = qE$. The electric force on a positive charge in an electric field is in the direction of E, while; The electric force on a negative charge is opposite to the direction of E.; Hence a +ve charge placed in an ...

Electric Fields - A-Level Physics Revision - XtremePapers ...

In my experience, red phosphorus is less commonly used in labs at this level (it isn't as excitingly reactive as white phosphorus!) - which is why I am concentrating on the white form. Sulphur If a stream of chlorine is passed over some heated sulphur, it reacts to form an orange, evil-smelling liquid, disulphur dichloride, S₂Cl₂.

chemical reactions of period 3 elements - A-Level ...

A-Level: Chemistry: Inorganic Chemistry: Periodic Table Group 2: THERMAL STABILITY OF THE GROUP 2 CARBONATES AND NITRATES. This page looks at the effect of heat on the carbonates and nitrates of the Group 2 elements - beryllium, magnesium, calcium, strontium and barium. It describes and explains how the thermal stability of the compounds ...

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