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$\text{SO}_2(\text{g}) + \text{O}_3(\text{g}) \rightarrow \text{SO}(\text{g}) + \text{O}(\text{g})$ $\text{SO}_2(\text{g}) + \text{H}_2\text{O}_2(\text{l}) \rightarrow \text{H}_2\text{SO}_4(\text{aq})$ (b) Oxides of Nitrogen: Dinitrogen and dioxygen are the main constituents of air. These gases do not react with each other at a normal temperature. At high altitudes when lightning strikes, they combine to form oxides of nitrogen.

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