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Page 1/33

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photon is calculated by combining Einstein 's equation for

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radiation The physical reality underlying de Broglie waves is a subject of ongoing debate. Some theories treat either

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In 1924, Louis de Broglie proposed a new speculative hypothesis that electrons and other particles of matter can behave like waves.

Page 11/33

Today, this idea is known as de Broglie's hypothesis of matter waves. In 1926, De Broglie's hypothesis, together with Bohr's early quantum theory, led to the development of a new theory of wave quantum mechanics to describe the physics of atoms and subatomic particles.

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the most obvious — and the most principled defence is ...

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V" that to each portion of energy with a proper mass m 0 one may associate a periodic phenomenon of the frequency 0, such that one finds: $h = 0 = m \cdot 0 c$ 2. The frequency 0 is to be measured, of course, in the rest frame of the energy packet ...

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beam of electrons can also be diffracted just like a beam of light. In essence, the de Broglie equation helps us understand the idea of matter having a wavelength.

de Broglie Equation -Derivation, de Broglie Wavelength ... Abstract: Introductory course on quantum and Page 25/33

atomic physics including optics and statistical physics. Objective: A basic introduction to quantum and atomic physics, including basics of optics and equilibrium statistical physics.

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Broglie wave phenomena
of larger and more
Page 26/33

complex objects than atoms, fullerenes come to mind as suitable candidates. After their discovery9 and the subsequent invention of ef fi cient massproduction methods 10. they became easily available. In our experiment (see Fig. 1) we use commercial, 99.5% pure, C

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De Broglie waves account for the appearance of subatomic particles at conventionally unexpected sites because their waves penetrate

Page 29/33

barriers much as sound passes through walls. Thus a heavy atomic nucleus occasionally can eject a piece of itself in a process called alpha decay. The piece of nucleus (alpha particle) has insufficient energy as a particle to overcome the force barrier surrounding the nucleus; but as a wave it can leak through the barrier—that Page 30/33

is, it has a finite Z probability ...

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