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Report Sheet - Lab 7 D.2 Formulas of Ionic compounds and
Førncus Formula PO calcium ion ironO Negaøve Ion co 2- N
03 — 03 — OH - more Name Potassium carbonate Sodium
nitrate Calcium bicarbonate Aluminum h droxide Lithium ho

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Write Potassium sulfate Positive Ion 24 D.3 Names of ionic compounds CaSO_4 $\text{Al}(\text{NO}_3)_3$ Na_2CO_3 MgSO_4 $\text{Cu}(\text{OH})_2$ Calcium sulfate

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Compounds and Their Formulas Report Sheet Lab - 7 B.1

Physical properties Compound Appearance Density Melting

Point B.2 Formulas of Ionic compounds Name Positive Ion

Negative Ion Formula Sodium chloride Na^+ Cl^- Magnesium

chloride Calcium oxide Lithium phosphide Aluminum sulfide

Calcium nitride B.3 Names of Ionic compounds K_2S

Potassium sulfide BaF_2 MgO Na_3N AlCl_3 Mg_3P_2 C.

Ionic Compounds with Transition Metals C.1 Physical

properties

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Compound name : Molecular weight: Molecular formula: 1:
Acetic acid: 60.052 g/mol: CH₃COOH: 2: Hydrochloric acid:

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36.458 g/mol: HCl: 3: Sulfuric acid 98.072 g/mol: H₂SO₄: 4: Acetate: 59.044 g/mol: CH₃COO⁻: 5: Ammonia: 17.031 g/mol: NH₃: 6: Nitric acid: 63.012 g/mol: HNO₃: 7: Phosphoric acid: 97.994 g/mol: H₃PO₄: 8: Sodium phosphate: 119.976 g/mol: Na₃PO₄: 9: Calcium carbonate: 100.086 g/mol: CaCO₃: 10: Ammonium sulfate

Chemical Compound Formula | Formula Chart

lab 7 - Compounds and Their Formulas Report Sheet Lab 7 ...

Question: Report Sheet-Lab 7 D.2 Formulas Of Ionic

Compounds Formula Positive Ion Negative Ion Name

Potassium Carbonate K Sodium Nitrate Calcium Bicarbonate

Aluminum Hydroxide Lithium Phosphate Potassium Sulfate

Co₂- D.3 Names Of Ionic Compounds CaSO₄ AlNO₃ Na CO

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M&SO

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