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4NH₃(g) + 5O₂(g) → 4NO(g) + 6H₂O(g)

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kJ mol^{-1} 2 $2+$ $+$ $=$ $-$ HÖ 910 -1

Describe the effect, if any, of each of the following changes on the equilibrium concentration of nitrogen monoxide in a particular equilibrium mixture, giving a reason in each case. (i) Increasing the pressure, at constant

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temperature [2]

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– 2 – Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student

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determined the value of x in the formula of hydrated ethanedioic acid, $\text{HOOC} - \text{COOH} \cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of $\text{NaOH} (\text{aq})$. 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a

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total ...

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Ethane, C_2H_6 , and disilane, Si_2H_6 ,
are both hydrides of group 4
elements with similar structures
but with different chemical

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properties. (a) (i) Deduce the Lewis (electron dot) structure for Si_2H_6 showing all valence electrons. [1] (ii) State and explain the $\text{H}-\text{Si}-\text{H}$ bond angle in Si_2H_6 . [2]

CHEMISTRY Candidate session

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number STANDARD LEVEL
PAPER 2

[2] (b) Marit arranged the values
she found in Table 12 into an

energy cycle. $C_2H_4(g) + H_2(g)$
 $C_2H_6(g)$ 3 0 2 $2CO_2(g) + 3H_2O$ –
1 4 1 1 k J m o l – 1 – 2 8 6 k J m
o l – 1 – 1 5 6 0 k J m o l – 1

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(hydrogenation) (I) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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