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chemistry lessons of
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chemical properties.
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(Physical & Chemical)
3.3 Mixtures of
Matter (Homo &
Hetero) 3.4 Pure
Substances (Elements
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Vocabulary. STUDY.

PLAY. Chemical

Change. A process
involving one or

more substances

changing into a new

substance. Metal. An

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element that is a solid at room temperature, a good conductor of heat and electricity, and is generally shiny. Solution.

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change to a sample
of matter in which

some properties of
the material change,
but the identity of
the matter does not.

8.5: Changes in
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Atkins, de Paula &
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Formulas of Hydrates

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6. From a laboratory
process designed to
sepa-rate water into
hydrogen and
oxygen gas, a student

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collected 10.0 g of hydrogen and 79.4 g of oxygen. How much water was originally involved in the process? mass reactants mass products mass reactants mass water electrolyzed

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stuff of which the
universe is
composed. All matter

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has two characteristics: Mass;
Occupies space;
States of Matter-The
distinct forms that
different phases of
matter take on -
Solid- rigid, has a
fixed shape and
volume; Ex. ICE -
Liquid- has a definite
volume and takes the
shape of the
container; Ex. WATER

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