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A change in physical form of a substance or physical properties of a substance like color, melting point, state, size, shape, and arrangement of molecules is known as physical change. A change in chemical properties of a substance

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like composition, reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be formed.

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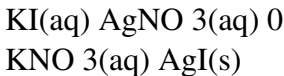
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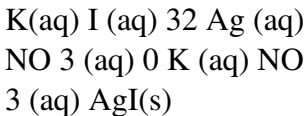
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chemical equation:



complete ionic equation:



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Solution. Physical:

boiling and melting are
physical changes. When
water boils, no bonds
are broken or formed.

The change could be
written: $\text{H}_2\text{O}(l) \rightarrow \text{H}_2\text{O}(g)$

Chemical: The dark

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grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

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3.6: Changes in Matter:
Physical and Chemical
Changes ...

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CHAPTER 12

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MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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