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Three States of Matter\" The
Creation of Chemistry - The

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and Measurement States of matter
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nature of matter - Part-1 -
Chemistry Changes of States of
Matter Chemical and Physical

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States of Matter : Solid Liquid Gas
10 Amazing Experiments with
Water

Pure Substances and Mixtures!
(Classification of Matter) Physical
and Chemical Changes

Understanding Atoms, elements,

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Basic Chemistry Series - 1.2 What
is matter? - HD Properties of
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- States of Matter \u0026

Changing State #20 Pure

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Elements \u0026 Compounds,

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Classification of Matter, Chemistry
Examples.

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Matter and Change. Ch 1 –

Introduction to Chemistry 1.1 The

Stories of Two Chemicals Ozone

Layer, atmosphere, ozone

formation, chlorofluorocarbons,

CFC ' s 1.2 Chemistry and Matter

Chemistry Central Science

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Chapter Summaries – Chemistry
Matter and Change

Chapter 1 Section 2 MATTER and
Its Properties Matter – Anything
that has MASS and VOLUME.

ATOMS and MOLECULES the
basic building blocks of matter.

ATOM-- the smallest unit of an

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element that still maintains the
properties that element.

ELEMENT – a pure substance
made up of only one type of atom.

MATTER and CHANGE Chapter 1
Section 1

Chemistry: Matter and Change

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Chapter 6 Study Guide for Content
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Class STUDY GUIDE FOR
CONTENT MASTERY Section 6.3
Periodic Trends In your textbook,
read about atomic radius and ionic
radius. Circle the letter of the
choice that best completes the

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statement or answers the question.

1. Atomic radii cannot be
measured ...

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CHAPTER 3 SOLUTIONS

MANUAL Matter—Properties and
Changes Matter—Properties and
Changes Solutions Manual

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Chapter 3 35 Section 3.1

Properties of Matter pages 70 – 75

Problem-Solving Lab 1. Explain

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Study the flow of a compressed gas must be controlled for practical and safe use.

Matter—Properties and
Changes Matter—Properties and
Changes

Chapter 1 VIDEODISC Cosmic

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Effect, Movie Ozone Studies, Movie
Nylon, Movie Dr. Jekyll and Mr.
Hyde, Movie Lab Safety, Movie CD-
ROM Chemistry: Matter and
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Chemistry, Demonstration The
following multimedia for this

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chapter are available from
Glencoe.

Chapter 1: Introduction to
Chemistry

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Teacher Guide and Answers 7

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Periodic Table and Periodic Law

Section 6.1 Development of the
Modern Periodic Table 1. octaves
2. eight 3. nine 4. accepted 5.
Dmitri Mendeleev 6. atomic mass
7. elements 8. atomic number 9.
Henry Moseley 10. protons 11.
periodic law 12. properties 13. 14

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Ch 6 Study Guide answers

Date Class Chapter Assessment

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Chapter 5 25 Electrons in Atoms

Reviewing Vocabulary Match the
definition in Column A with the

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term in Column B. Column A

Column B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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Chapter Assessment Content

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key. some numbers used in

chemistry can be q... the study of

energy changes (that is, energy

produced or absor... A system that

can exchange neither energy nor

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Change. extensive property.
intensive property. extensive

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properties (examples) intensive
property (examples) a property
that depends on the amount of
matter in a sample. a property that
depends on the type of matter in a
sample not.... mass, volume,
length, weight.

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chapter 2 Flashcards and Study ...

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Chapter 11 31 number of moles of
known substance C H C H C H O O
O CO CO CO H O H O H O number
of moles of unknown substance O

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CO H O C H CO H O C H O H O C
H O CO use mole ratio 5 mol O 2 1
mol C 3 H 8 3 mol CO 2 1 mol C 3
H 8 4 mol H 2 O 1 mol C 3 H 8 1
mol C 3 H 8 5 mol O 2

MATH SKILLS TRANSPARENCY
MASTER 16

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A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a

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chemical change. Properties of
Compounds Quite different
properties than their component
elements.

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