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and Change part
1 (Chem 1 H)

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Matter\ " The
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—Introduction:

Matter and

Measurement

States of matter

| States of

matter and

intermolecular

forces |

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12 - Matter and
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nature of matter

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Change Chapter

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Changes of

States of Matter

Chemical and

Physical Changes

States of Matter

: Solid Liquid

Gas10 *Amazing*

Experiments with

Water

Pure Substances

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Matter And Mixtures!

(Classification
of Matter)

Physical and

Chemical Changes

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Understanding
Atoms, elements,
and molecules

Part #1 (9min)

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What is matter?

- HD Properties

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of Matter And

Physical and
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Changes:

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Changes States

of Matter—

Solids, Liquids,

Gases \u0026

Plasma—

Chemistry Change

of State |

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States of Matter

and Changes of

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Chemistry -

States of Matter

\u0026 Changing

State #20 Pure

Substances and

Mixtures,

Elements \u0026

Compounds,

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Classification
of Matter,
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Examples,

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Classifying
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Chemistry and
Matter Chemistry
Central Science

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Chapter 1
Section 2 MATTER
and Its

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Properties of Matter

— Anything that has MASS and VOLUME. ATOMS and MOLECULES are the basic building blocks of matter.

ATOM- - the smallest unit of an element that still maintains the properties of that element.

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ELEMENT—a pure substance made up of only one type of atom.

Stoichiometry

MATTER and CHANGE Chapter 1 Section 1

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Solutions Manual

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3 35 Section 3.1

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Properties of
Matter pages
70–75 Problem-
Solving Lab 1.
Explain why the
flow of a
compressed gas
must be
controlled for
practical and
safe use.

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Table and

Periodic Law

Section 6.1

Development of
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Periodic Table

1. octaves 2.
eight 3. nine 4.

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accepted 5.
Dmitri Mendeleev
6. atomic mass
7. elements 8.
atomic number 9.
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properties 13.
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5 25 Electrons

in Atoms

Reviewing

Vocabulary Match

the definition

in Column A with

the term in

Column B. Column

A Column B 1.

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The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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extensive
property.

intensive
property.

extensive
properties

(examples)

intensive
property

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(examples) a
property that
depends on the
amount of matter
in a sample. a
property that
depends on the
type of matter
in a sample
not... mass,
volume, length,
weight.

chemistry matter

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11 31 number of

moles of known

substance C H C

H C H O O O CO

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CO₂ H₂O H₂O H₂O
 0 number of
 moles of unknown
 substance 0 CO₂ H₂O
 0 C₂H₆ CO₂ H₂O C₂H₆
 0 H₂O C₂H₆ 0 CO₂
 use mole ratio 5
 mol 0 2 1 mol C
 3 H₂ 8 3 mol CO₂ 2
 1 mol C₂H₆ 3 H₂ 8 4
 mol H₂ 2 0 1 mol
 C₂H₆ 3 H₂ 8 1 mol C
 3 H₂ 8 5 mol 0 2

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A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances

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are converted
into different
substances. Heat
and light are
often evidence
of a chemical
change.

Properties of
Compounds Quite
different
properties than
their component
elements.

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