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chemistry and chemical calculations |
Brief answer 44 - 1 **Chemistry 110 --**
Chapter 11, Part Two: Unsaturated
Hydrocarbons *Chemistry Chapter 11*
Answers

11. molar mass = mass/moles; mass =
45.0 g; find moles; $T_b = 90.2^\circ\text{C} -$
 $80.1^\circ\text{C} = 10.1^\circ\text{C}$

$\Delta T_f = 2.53^\circ\text{C}$
 $\Delta T_f = K_f \cdot m$
 $2.53 = 5.12 \cdot m$
 $m = \frac{2.53}{5.12} = 0.494$
 $m = \frac{\text{mass}}{\text{moles}}$
 $0.494 = \frac{\text{mass}}{0.0750}$
 $\text{mass} = 0.494 \cdot 0.0750 = 0.03705$
 $\text{mass} = 0.03705 \cdot 1000 = 37.05$
 $\text{mass} = 37.05 \text{ g}$

Answers to Chapter 11 Study

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Questions - Chemistry LibreTexts
NCERT Book Chapter 1. NCERT Answers Chemistry Part - 1. Chapter 11: the p - block elements. The branch of science under which the structure of matter, its properties, uses and interaction with other types of a matter and various forms of energy are studied is called chemistry. <https://www.tiwariacademy.com/ncert-solutions/class-11/chemistry/>

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SOLVED. 11.1. Classify the following
as primary, secondary and tertiary

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alcohols. Ans: Primary alcohols: (i), (ii), (iii) Secondary alcohols: (iv), (v) Tertiary alcohols: (vi) 11.2. Identify allylic alcohols in the above examples. Ans: (ii) and (iv) i.e. $\text{H}_2\text{C}=\text{CH}-\text{CH}_2\text{OH}$ and.

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Question 11. What would be the reactant and reagent used to obtain 2, 4-dimethyl pentan-3-ol? (a) Propanal and propyl magnesium bromide (b) 3-methylbutanal and 2-methyl magnesium iodide (c) 2-dimethylpropanone and methyl magnesium iodide (d) 2-methylpropanal and isopropyl magnesium iodide. Answer. Answer: (d) 2-methylpropanal and isopropyl magnesium iodide

*MCQ Questions for Class 12
Chemistry Chapter 11 Alcohols ...*

Answer: The basic theme of
organisation of elements in the

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periodic table is to simplify and systematize the study of the properties of all the elements and millions of their compounds. This has made the study simple because the properties of elements are now studied in form of groups rather than individually.

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