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Sample Problem #2. Write the chemical equation for the following reaction:

Hydrochloric acid and solid sodium hydrogen carbonate react to produce aqueous sodium chloride, water, and carbon dioxide. Hint: Acids will always be aqueous unless otherwise stated. $\text{HCl (aq)} + \text{NaHCO}_3(\text{s}) \rightarrow \text{NaCl (aq)} + \text{H}_2\text{O (l)} + \text{CO}_2(\text{g})$

~~Chapter 11: Chemical Reactions~~

Chemical Reactions Guided Practice Problems Chapter 11 Chemical Reactions 119 GUIDED PRACTICE PROBLEM 2 (page 324) 2. Sulfur burns in

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oxygen to form sulfur dioxide. Write a skeleton equation for this chemical reaction. Include appropriate symbols from table 11.1. Analyze Step 1. Identify the

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Chemical Reactions Guided Practice Problem CHEMICAL REACTIONS (pages 330-339). This section GUIDED PRACTICE PROBLEM 2 (page 324). 2. possibly a change of state. Essential Questions and Answers. 1. What are the types for energy? Chemical Energy is energy caused by chemical reactions. Quiz: Chemical Reactions. 1/8.

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During chemical reactions, atoms.

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Problems Answers chapter 11 Guided Reading.

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Balancing Equations: Answers to Practice Problems 1.

Balanced equations.

(Coefficients equal to one (1) do not need to be shown in your answers).

~~Balancing Equations: Practice Problems~~

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Practice: Balancing chemical equations 1. This is the currently selected item. Next lesson. Stoichiometry.

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13) Using the equation from

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~~Problem #12, determine the mass of aluminum acetate that can be made if I do this reaction with 125 grams of acetic acid and 275 grams of aluminum hydroxide. 14) What is the limiting reagent in problem #13? 15) How much of the excess reagent will be left over in problem #13 after the reaction is complete?~~

~~Stoichiometry Practice Worksheet~~

Name: _____ Unit 11 Honors Chemistry Notes

Thermodynamics - the study of _____. Thermochemistry is the study of how heat changes during phase changes and chemical reactions.

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Quantifying Energy (ability to do work or produce heat)
Calories Joules Law of Conservation of Energy - 1st Law of Thermodynamics ex) Potential Energy stored in gasoline bonds ? Kinetic energy in ...

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Stoichiometry Practice

Worksheet Solve the

following stoichiometry

grams-grams problems: 1)

Using the following

equation: $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2$

$\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$ How many

grams of sodium sulfate will

be formed if you start with

200.0 grams of sodium

hydroxide and you have an

excess of sulfuric acid? 2)

Using the following

equation:

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