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$3e \text{ N V RT} \times 18 \text{ mol}$

$\times 8.314 \times 10^{-5}$

(bar \square m³ mol K)

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$T_{P=1} = VV_1 \times$

$10^{-3} - (10 - x)$

106 volume taken
up by liquid

Equating P and P_{vap} we have x

$\times 18$

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$P = 5432.8 \times l$

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