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~~Bonding lecture 1 of~~

~~3 Chapter 8 Covalent~~

~~Bonding Answer~~

When H forms a bond

with H O to form the

hydronium ion H O ,

this bond is called a

coordinate covalent

bond because a. both

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Answer Key

bonding electrons come from the oxygen atom. b. it forms an especially strong bond. c. the electrons are equally shared. d. the oxygen no longer has eight valence electrons.

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mixing of several atomic orbitals to form the same total number of equivalent hybrid orbitals.

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Section 8.1 The

Covalent Bond pages

240–247 Practice

Problems page 244

Draw the Lewis

structure for each

molecule. 1. PH₃

HH H—HP

respectively, for

single, double, and

triple P— — 2. H₂S

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HHH—HSS ...

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COVALENT BONDING

Class 8.2 8.2 8.2 8.2

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Chapter Quiz Choose

the best answer and

write its letter on the

line. 1. A bond in

which each atom

contributes two

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electrons is a. a double covalent bond. b. an ionic bond. c. a polar covalent bond. d. a coordinate covalent bond. 2. The electron dot structure for hydrogen sulfide, H₂S ...

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question When
sharing of electrons
occurs the

attachment between
atoms is called
answer covalent bond
question in a covalent
bond, the
dissociation energy is

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Covalent Bonding Or
How I Learned to
Love Sharing (But
Remember, File
Sharing is Illegal) 2.

As you should
remember, ionic
compounds are solids
at room
temperatures that
have one ion strip the

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electron(s) from the
other elements '
electron cloud.

Answer Key

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Bonding and

Molecular Structure

8-11. nuclei. This

results in stronger

attractive forces

between electrons

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and nuclei,

decreasing the distance between the nuclei. A carbon-carbon single bond has a bond order of 1 and is longer than a carbon-carbon double bond with a bond order of 2.

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takes many forms in
nature: In this

chapter, we are going

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to learn to

distinguish the type of compound that we have already studied,

the “ ionic compound ” (which contains oppositely-charged particles: metal cations and non-metal anions), from a different type of compound – a “ molecular compound ” .

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Quiz Answer Section

MULTIPLE CHOICE 1.

ANS: D PTS: 1 DIF: L3

REF: p. 248 | p. 249

OBJ: 8.4.1 Describe

how

electronegativity

values determine the

charge distribution in

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a polar molecule.

MSC: application

2.ANS: B PTS: 1 DIF:

L1 REF: p. 251

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