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Compounds; Key Terms; Summary; Exercises

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(a) Sc: $[\text{Ar}]4s^2 3d^1$; (b) Ti: $[\text{Ar}]4s^2 3d^2$; (c) Cr:
 $[\text{Ar}]4s^1 3d^5$; (d) Fe: $[\text{Ar}]4s^2 3d^6$; (e) Ru: $[\text{Kr}]5s^2$
 $4d^6$. 3. (a) La: $[\text{Xe}]6s^2 5d^1$, La $3+$: $[\text{Xe}]$; (b) Sm:
 $[\text{Xe}]6s^2 4f^6$, Sm $3+$: $[\text{Xe}]4f^5$; (c) Lu: $[\text{Xe}]6s^2 4f^{14}$
 $5d^1$, Lu $3+$: $[\text{Xe}]4f^{14}$. 5. Al is used because it is the
strongest reducing agent and the only option listed that
can provide sufficient driving force to convert La (III)
into La.

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if large amounts of acid or base are added. A buffer
system should contain considerably more acid than
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