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GCSE Chemistry - Formation of Ions

~~#11 What are Ions? Atoms, Ions, and Isotopes (including Atomic Notation)~~

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The atomic symbol for an ion, it has the mass number, atomic number and

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the amount of electrons added or subtracted What is the difference between the atomic mass and the atom number The mass number is the protons and neutrons added together and the atomic number is the mass of all isotopes of that element.

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