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In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and

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would, therefore, have
centripetal acceleration in
order to orbit the nucleus.

Atomic Structure: Chapter
Problems

Question 6. Define the

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following terms: (a) Atomic number (b) Mass number (c) Nucleons (d) Valence shell

Answer: (a) Atomic number: Atomic number refers to the number of protons present in an atom. It is denoted by Z . Example: An atom of oxygen

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contains 8 proton Therefore
its atomic number is 8.

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each question in order to

receive credit (1) Indicate

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whether each of the following is a permissible set of quantum numbers. If the set is not permissible, state why it is not a.

2, 4 0 $m = +1$ b. $n = 5, l = 3, m = -3$ c. $n = 4, l = 1, m = 3$

d. $n = 0, l = 1, m = 0$ e. $n = 4, l = 1, m = 4$

(2) Indicate whether each of

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which includes, questions on
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