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Atomic Structure Worksheet – Solution Key. Fill in the blanks for the  
elements in this chart. For the purposes of this chart, round all atomic  
masses to the nearest whole number. Element Number of Protons  
Number of Neutrons Number of Electrons Atomic Mass Atomic  
Number. lithium 3 4 3 7 3.

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www.njctl.org Chemistry Atomic Structure Answers 1. In the nuclear

model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.

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Atomic Structure: Chapter Problems

Question 6. Define the following terms: (a) Atomic number (b) Mass number (c) Nucleons (d) Valence shell Answer: (a) Atomic number:

Atomic number refers to the number of protons present in an atom. It is denoted by  $Z$ . Example: An atom of oxygen contains 8 proton

Therefore its atomic number is 8.

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Name Datu Directions: read and answer each question in order to receive credit (1) Indicate whether each of the following is a permissible set of quantum numbers. If the set is not permissible, state why it is not a.  $2, 4, 0, m = +1$  b.  $n = 5, 3, m = -3$  c.  $n = 4, l = 3, m = 2$  d.  $n = 0, l = 0, m = 4 + 1$  (2) Indicate whether each of the following is a permissible set of quantum numbers.

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