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MARK SCHEME - GCSE SCIENCE A / CHEMISTRY - CH1HP - JUNE 2014 6 of 14
Question Answers Extra information Mark AO / Spec. Ref. ID 1(a)(i) 2,4 drawn (as dots / crosses / e-) 1 2 / 1.1.1h E 1(a)(ii) Water (vapour) / steam allow hydrogen oxide / H 2 O do not accept hydroxide 1 1 / 1.4.3b G 1(b) any two pairs from:
carbon dioxide (1)

SCIENCE A / CHEMISTRY

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Information to Examiners 1. General The mark scheme for each question shows: the marks available for each part of the question the total marks available for the question the typical answer or answers which are expected extra information to help the Examiner make his or her judgement and help to delineate what is ...

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G/Jun14/CH1HP Answer all questions in the spaces provided. 1 Fossil fuels contain carbon and hydrogen. 1 (a) (i) Use the Chemistry Data Sheet to help you to answer this question. Complete Figure 1 to show the electronic structure of a carbon atom. [1 mark] Figure 1 1 (a) (ii) Complete the word equation for the oxidation of hydrogen. [1 mark]

~~Science A CH1HP Unit Chemistry C1 H~~

Aqa Science Ch1hp Answers 2014 MARK SCHEME – GCSE SCIENCE A / CHEMISTRY – CH1HP – JUNE 2014 9 of 14 Question Answers Extra information Mark AO / Spec. Ref. ID 3(a)(i) add bromine (water) to the oil one drop at a time / dropwise or count / record number of drops until the bromine (water) no longer decolourises or until

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Question Answers Extra information Mark AO / Spec. Ref. ID 3(a)(i) any one from:
destruction of habitats fewer plants/trees to absorb carbon dioxide dust / noise / air
/ visual (pollution) more traffic allow breathing problems 1 3 / 1.2 E

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AQA: June 2014 (4405) Science A - Unit 1 Biology B1 Higher (BL1HP) Q A: AQA:

June 2014 (4405) Science A - Unit 1 Chemistry C1 Foundation (CH1FP) Q A: AQA:

June 2014 (4405) Science A - Unit 1 Chemistry C1 Higher (CH1HP) Q A: AQA: June

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Accept / allow Accept is used to indicate an equivalent answer to that given on the left-hand side of the mark scheme. Allow is used to denote lower-level responses that just gain credit.

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3 Do not write outside the box (03) G/Jun17/CH1HP Turn over Answer all questions in the spaces provided. 1 Elements are made of atoms. 1 (a) Table 1 shows the atomic numbers and mass numbers of three atoms. Table 1 Atom Atomic number Mass number 1 12 24 2 12 25 3 12 26 1 (a) (i) Suggest, in terms of the number of subatomic particles, why the atomic numbers of the

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