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experiment.  $\text{Ni}(\text{NO}_3)_2 + \text{H}_2\text{O}$ .  $\text{Ni}^{2+}$   
reacts with concentrated ammonia s....  
 $\text{Ni}^{2+}$  reacts with ethylene diamine  
 $\text{Ni}^{2+}$ ....  $\text{Cu}(\text{OH})_2 + 2 \text{NH}_3 + 2 \text{NH}_4^+$ .  
 $[\text{Ni}(\text{H}_2\text{O})_6]^{2+}$ ...  $\text{Ni}^{2+} + 2 \text{NO}_3^-$ ...  
hexaaquanickel (II)... green coloured....

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30  $\mu$ l aliquot of the reaction mixture and  
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Experiments General Chemistry Lab

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following laboratories have been compiled  
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12: Making Soap - Saponification

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Experiments ... When you have completed the experiment, dissolve all your heated residues in water, put all your solids and liquids in the waste crock. Pre-laboratory Assignment: Properties of Hydrates.

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regarding its electronic and mo;

3. Suspension bridge are known to be normal around the world especially in developed countries. These b; 4.

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4. Milk has a density of 1.04 g/mL. What is the mass in grams of 1.00 quart of milk? a. 984 g b. 0.98; 5.

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