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One of the chambers has volume V 1 V 1 and contains ideal gas at pressure P 1 P 1 and temperature T 2 T 2. If the partition is removed without doing any work on the gas, the final equilibrium temperature of the gas in the container will be. A.

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There are three parts in the question paper. The distribution of marks subjectwise in each correct response. Part A

CHEMISTRY (144 marks)

Question No. 1 to 24 consist FOUR (4) marks each and Question No. 25 to 30 consist EIGHT (8) marks each for each correct response.

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D. CH3Cl. CHECK ANSWER. Explanation. $S_{S_N}^2$ reaction is favoured by small groups on the carbon atom attached to halogen. So, the order of reactivity is. $C_{H_3}^2$ reaction is favoured by small groups on the carbon atom attached to halogen. So, the order of reactivity is. $C_{H_3}^2$ reaction is favoured by small groups on the carbon atom attached to halogen. So, the order of reactivity is. $C_{H_3}^2$ reaction is favoured by small groups on the carbon atom attached to halogen. So, the order of reactivity is.

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