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NUCLEAR CHEMISTRY 701 SECTION 22-1 OBJECTIVES Explain what a nuclide is, and

describe the different ways nuclides can be represented Define and relate the terms mass defect and

nuclear binding energy 22 1 Nuclear

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Chemistry Section 22-1: The Nucleus • Atomic nuclei= protons and neutrons (together are nucleons) o

Nuclide= an

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22.1-2 Beta emission (occurs when an atom has too many neutrons, or not enough protons). In the

nucleus a neutron splits into an electron and a proton ($1\ 0\ n \rightarrow 1\ 1\ p + 0\ -1\ e$) Since this electron

originated from the nucleus, it is a beta particle.

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Emission of α -particle from $^{92}_{238}\text{U}$. Ans- $^{92}_{238}\text{U} \rightarrow \alpha + ^{90}_{234}\text{Th}$. b) Emission of positron β^+ from $^{14}_{27}\text{Si}$.

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